

Complexes with the Di- μ -alkoxy-tetrakisphenyldiazenido-dimolybdenum Core, $[\text{Mo}_2(\text{NNPh})_4(\text{OR})_2]^{2+}$. The Synthesis and Structural Characterization of the Phenylhydrazino-derivative, $[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_4(\text{H}_2\text{NNHPh})_2]$

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The organodiazenido-group, $-\text{NNR}$, is of intrinsic chemical interest as a model for understanding the chemistry of metal-bound dinitrogen and NO [1]. The organodiazenido-group is known to form robust complexes with molybdenum, often displacing oxo-groups to give the *cis*- $[\text{Mo}(\text{NNR})_2]^{2+}$ structural unit [2]. We have recently extended this chemistry to the polyoxomolybdates with the characterization of species of the type $[\text{Mo}_4\text{O}_8(\text{OR})_2(\text{NNR})_4]^{2-}$ [3]. At the time, we speculated on the potential utility of the $[\text{Mo}_2(\text{NNR})_4(\text{OR})_2]^{2+}$ unit of this species as a core for further synthetic and structural studies. Immediately thereafter, details of the synthesis and structural characterization of $[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_2(\text{acac})_2]$ were reported [4]. Although the synthetic approach to this latter complex varied from that employed in our studies, it did suggest the potential for varied chemistry for the $[\text{Mo}_2(\text{NNPh})_4(\text{OR})_2]^{2+}$ unit. We report herein the synthesis and structural characterization of $[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_4(\text{H}_2\text{NNHPh})_2]$, a species containing this core and exhibiting molybdenum-hydrazino coordination, to our knowledge the first example of a molybdenum-diazenido-hydrazino complex.

Reaction of $(\text{Bu}^n)_2\text{N}_2[\text{Mo}_4\text{O}_8(\text{OMe})_2\text{NNPh}_4]$ with phenylhydrazine in methanol yields lustrous black crystals of $[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_4(\text{H}_2\text{NNHPh})_2]$. The infrared spectrum shows bands at 1605 and 1565 cm^{-1} characteristic of $\nu(\text{N}-\text{N})$ for *cis* coordination of the diazenido groups. The band at 2815 cm^{-1} is associated with $\nu(\text{C}-\text{H})$ for the bridging methoxy-groups and the medium intensity bands at 3210 and 3385 cm^{-1} are assigned for $\nu(\text{N}-\text{H})$ for the coordinated H_2NNHPh ligand. The complex may also be synthesized by reaction of $[\text{MoO}_2(2,3\text{-butanediolate})_2] \cdot 2\text{butane-2,3-diol}$ [5] with excess phenylhydrazine in methanol. Analogous reactions in ethanol/chloroform yield the species $[\text{Mo}_2(\text{NNPh})_4(\text{OEt})_2\text{Cl}_2(\text{HOEt})_2]$ [6].

Crystal Data: $[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_4(\text{H}_2\text{NNHPh})_2]$, monoclinic space group $P2_1/n$, $a = 11.437(2)$, $b =$

$12.004(2)$, $c = 16.756(3)$ Å, $\beta = 108.45(1)^\circ$, $V = 2182.2(6)$ Å³, $Z = 2$ to give $D_c = 1.45$ g cm⁻³. Structure solution and refinement based on 2009 reflections with $I_o \geq 3\sigma I_o$ converged at $R = 0.047$, with a goodness of fit of 1.657.

The structure of $[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_4(\text{H}_2\text{NNHPh})_2]$, illustrated in Fig. 1 consists of binuclear molecular units lying on a center of symmetry, with two chemically and crystallographically related $[\text{Mo}(\text{NNPh})_2(\text{OCH}_3)(\text{H}_2\text{NNHPh})]$ moieties linked by two bridging methoxy-groups. The coordination geometry about the molybdenum is distorted octahedral, generated by ligation to the α -N donors of two phenyldiazenido units, to the two bridging methoxy and the one terminal methoxy groups, and to the α -N of the phenylhydrazino-ligand, H_2NNHPh . The location of a center of symmetry on the mid-point of the $\text{Mo}-\text{O}2-\text{Mo}'$ vector imposes planarity on the $\text{Mo}-\text{O}2-\text{Mo}'$ rhombus. The entire $[\text{Mo}_2(\text{OCH}_3)_2(\text{NN-})_4]$ moiety of the central core is essentially planar, such that the complex $[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_4(\text{H}_2\text{NNHPh})_2]$ is structurally related to the $\{[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_2][\text{MoO}_4]^{2-}\}^{2-}$ precursors, that is $[\text{Mo}_4\text{O}_8(\text{OMe})_2(\text{NNPh})_4]^{2-}$, by substitution of the capping $[\text{MoO}_4]^{2-}$ units by two terminal methoxy groups and two phenylhydrazino-ligands.

The short $\text{Mo}-\text{N}(\text{diazenido})$ distances, 1.823(6) Å (average), together with the average $\text{N}-\text{N}$ distance of 1.243(9) Å and the essentially linearity of the $\text{Mo}-\text{N}-\text{N}$ unit for the two diazenido ligands, are

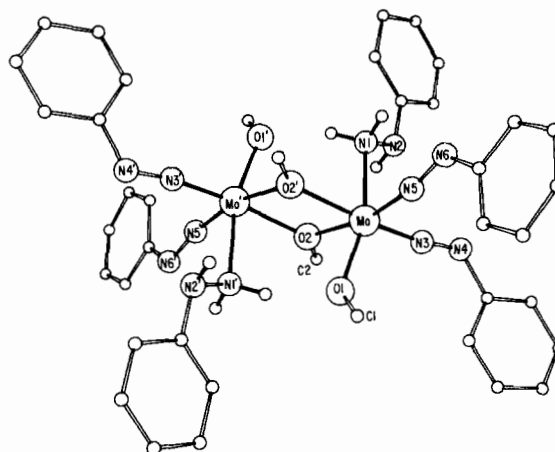


Fig. 1. ORTEP view of the structure of $[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_4(\text{H}_2\text{NNHPh})_2]$, showing the atom labelling scheme. Relevant bond lengths (Å) and angles ($^\circ$): $\text{Mo}-\text{O}1$, 1.990(5); $\text{Mo}-\text{O}2$, 2.110(4); $\text{Mo}-\text{N}1$, 2.265(5); $\text{Mo}-\text{N}3$, 1.816(6); $\text{Mo}-\text{N}5$, 1.829(6); $\text{Mo}-\text{O}2'$, 2.125(4); $\text{N}1-\text{N}2$, 1.445(9); $\text{N}3-\text{N}4$, 1.242(8); $\text{N}5-\text{N}6$, 1.243(9); $\text{N}1-\text{Mo}-\text{O}1$, 163.9(2); $\text{N}3-\text{Mo}-\text{O}2'$, 167.4(2); $\text{N}5-\text{Mo}-\text{O}2$, 166.8(2); $\text{O}2-\text{Mo}-\text{O}2'$, 72.9(2); $\text{Mo}-\text{N}1-\text{N}2$, 115.4(3); $\text{Mo}-\text{N}3-\text{N}4$, 170.9(5); $\text{Mo}-\text{N}5-\text{N}6$, 161.0(5).

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TABLE I. Comparison of Structural Parameters for Complexes Possessing the $[\text{Mo}_2(\text{NNPh})_4(\text{OR})_2]^{2+}$ Core, of the Types $[\text{Mo}_2(\text{NNPh})_4(\text{OR})_2\text{X}_2\text{L}_2]^{n-}$ and $[\text{Mo}_2(\text{NNPh})_4(\text{OR})_2(\text{LL})_2]^{n-}$ ^a

Complex	Mo–Mo (Å)	Mo–N(diazenido) (Å)	Mo–O(bridge) (Å)	Mo–X or Mo–L ^b (Å)
$[\{\text{Mo}_2(\text{NNPh})_4(\text{OMe})_2\}[\text{MoO}_4]_2]^{2-}$	3.465(2)	1.823(9) (ave.)	2.140(8) (ave.)	2.063(8) (O)
$[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_4(\text{H}_2\text{NNHPh})_2]$	3.406(1)	1.823(7) (ave.)	2.118(6) (ave.)	1.990(5) (O) 2.265(5) (N)
$[\text{Mo}_2(\text{NNPh})_4(\text{OEt})_2(\text{HOEt})_2\text{Cl}_2]$	3.407(1)	1.825(6) (ave.)	2.109(4)	2.440(2) (Cl) 2.148(5) (O) ^c
$[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_2(\text{acac})_2]$	3.296(3)	1.806(14) (ave.)	2.089(11)	2.055(8) (O) 2.167(7) (O)

^aX = monoanionic ligand; L = neutral ligand; LL = bidentate ligand. ^bX and L refer to fifth and sixth ligands coordinated to each Mo-center of the $[\text{Mo}_2(\text{NNPh})_4(\text{OR})_2]^{2+}$ core. Donor atom is identified in parenthesis. ^cRefers to oxygen donor of terminally coordinated ethanol ligand.

common to this grouping and indicative of extensive delocalization throughout the unit. The Mo–N1 (phenylhydrazine) distance of 2.265(5) Å is consistent with a molybdenum–nitrogen single bond, while the N1–N2 distance of 1.445(9) Å is similar to that observed for N–N single bonds in organohydrazines.

Although the $[\text{Mo}_2(\text{NNPh})_4(\text{OR})_2]^{2+}$ core is common to the structures of $[\text{Mo}_4\text{O}_8(\text{OR})_2(\text{NNPh})_4]^{2-}$, $[\text{Mo}_2(\text{NNPh})_4\text{Cl}_2(\text{OEt})_2(\text{HOEt})_2]$, $[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_4(\text{H}_2\text{NNHPh})_2]$ and $[\text{Mo}_2(\text{NNPh})_4(\text{OMe})_2(\text{acac})_2]$, two distinct structural forms are observed. While the first three examples exhibit a planar $[\text{Mo}_2(\text{OR})_2(\text{NN})_4]^{2+}$ moiety, the latter example presents a $[\text{Mo}_2(\text{OMe})_2(\text{NNPh})_4]^{2+}$ core with one phenyldiazenido-group on each Mo center rotated 90° to occupy a position axial to the plane of the $[\text{Mo}_2(\text{NNPh})_2(\text{OMe})_2]$ unit, in order to accommodate the terminally coordinated planar bidentate acac ligands. Table I compares bonding parameters for the four structures.

These observations suggest that the $[\text{Mo}_2(\text{NNPh})_4(\text{OR})_2]^{2+}$ unit provides a chemically robust and structurally flexible core, which may be prevalent in structures of the general types $[\text{Mo}_2(\text{NNPh})_4(\text{OR})_2\text{X}_2\text{L}_2]^{n-}$ and $[\text{Mo}_2(\text{NNPh})_4(\text{OR})_2(\text{LL})_2]^{n-}$ ($n = 0$ or -2 , depending on the nature of the coligands X, L or LL). In future publications, we will discuss the details of the syntheses and structural character-

istics of this growing class of complexes, representatives of a general type with chemistry potentially as extensive as that of the well-known μ -oxo-tetraoxodimolybdenum core, $[\text{Mo}_2\text{O}_5]^{2+}$.

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References

- 1 D. Sutton, *Chem. Soc. Rev.*, **4**, 443 (1975).
- 2 P. L. Dahlstrom, J. R. Dilworth, P. Shulman and J. Zubieta, *Inorg. Chem.*, **21**, 933 (1982); F. J. Lalor, D. Condon, G. Ferguson and M. A. Khan, *Inorg. Chem.*, **20**, 2178 (1981); G. Ferguson, B. L. Ruhl, F. J. Lalor and M. E. Deane, *J. Organomet. Chem.*, **286**, 75 (1985); T.-C. Hsieh and J. Zubieta, *Inorg. Chim. Acta*, **99**, L47 (1985); S. N. Shaikh and J. Zubieta, *Inorg. Chim. Acta*, **115**, L19 (1986).
- 3 T.-C. Hsieh and J. Zubieta, *Inorg. Chem.*, **24**, 1288 (1985); T.-C. Hsieh and J. Zubieta, *Polyhedron*, **5**, 305 (1986).
- 4 D. Carillo, P. Gouzerh and Y. Jeannin, *Nouv. J. Chim.*, **9**, 749 (1985).
- 5 R. J. Butcher, B. R. Penfold and E. Sinn, *J. Chem. Soc., Dalton Trans.*, 668 (1979).
- 6 T.-C. Hsieh and J. Zubieta, unpublished results.